

Writing Formulas Extra Practice!

- calcium chloride
- sulfur trioxide
- ammonium fluoride
- barium hydroxide
- aluminum hydrogen sulfate
- sodium iodate
- lithium bromide
- zinc sulfate
- iron (III) perchlorate
- nickel (II) permanganate
- silver hypochlorite
- carbon tetraiodide
- potassium dichromate
- dinitrogen trioxide
- copper (II) bromate
- lead (II) nitrate
- nitrogen trichloride
- ammonium carbonate
- manganese (IV) oxide
- tin (II) hydrogen carbonate
- silver chlorate
- chromium (III) acetate
- mercury (II) sulfite
- lead (IV) chlorite
- sodium nitrite
- beryllium phosphate
- magnesium phosphide
- iron (III) oxide
- chromium (II) sulfide
- oxygen difluoride
- copper (I) chromate
- sodium cyanide
- silicon tetrafluoride
- tin (II) fluoride

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Naming Compounds Extra Practice!

- CuBr
- $\text{Cr}_2(\text{SO}_3)_3$
- NiI_2
- Ba_3P_2
- NO_3
- FeO
- BeSO_4
- MnS
- K_2CO_3
- CS_2

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|-----|-------------------------------|-----|--|
| 6. | $\text{Pb}_3(\text{PO}_4)_2$ | 23. | $(\text{NH}_4)_2\text{Cr}_2\text{O}_7$ |
| 7. | MgSO_4 | 24. | PBr_5 |
| 8. | FeF_2 | 25. | KClO_3 |
| 9. | $\text{Al}_2(\text{CrO}_4)_3$ | 26. | CuHCO_3 |
| 10. | Cu_2CO_3 | 27. | $\text{Ni}_3(\text{PO}_4)_2$ |
| 11. | $\text{Zn}(\text{NO}_3)_2$ | 28. | AlBr_3 |
| 12. | AgClO_2 | 29. | CCl_4 |
| 13. | CO | 30. | K_3PO_4 |
| 14. | $\text{Sn}(\text{MnO}_4)_2$ | 31. | SO_3 |
| 15. | $\text{Ba}(\text{NO}_2)_2$ | 32. | N_2O_5 |
| 16. | N_2O | 33. | LiOH |
| 17. | NH_4NO_3 | 34. | $\text{Sr}(\text{ClO}_4)_2$ |

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