Problem Set #16

Name:

Part 1 - Balance the following nuclear equations and identify the type of nuclear reaction.

- 1. ${}^{239}_{94}Pu \rightarrow {}^{4}_{2}He + _$
- 2. ${}^{99}_{43}\text{Tc} \rightarrow ___ + {}^{0}_{-1}\text{e}$
- 3. ${}^{239}_{92}U + {}^{4}_{2}He \rightarrow ___ + {}^{1}_{0}n$
- 4. _____ + ${}^{1}_{0}n \rightarrow {}^{142}_{56}Ba + {}^{91}_{36}Kr + 3 {}^{1}_{0}n$
- 5. ${}^{1}_{0}n + {}^{235}_{92}U \rightarrow 2 {}^{1}_{0}n + ___ + {}^{137}_{52}Te$
- 6. ${}^{1}_{1}H + {}^{3}_{1}H \rightarrow ___$
- Part 2 Solve the following problems involving half-life.
- 7. Neptunium-239 has a half-life of 2.356 days. If 2.34375 grams of neptunium-239 remain after 11.78 days, how many grams of neptunium were in the original sample?
- 8. What is the half-life of hydrogen-3 (tritium) if 1.875 grams remains of a 30 gram sample after 49.28 years?
- 9. How long does it take a 160 gram sample of americium-241 to decay to 0.625 grams if the half-life of americium-241 is 432.2 years?
- 10. How much of a 125 gram sample of oxygen-22 remains after 13.5 seconds if the half-life of oxygen-22 is 2.25 seconds?